

**The 4<sup>th</sup> German-Japan Joint  
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**Presentation 13**

# **ORR Activity Enhancement of Carbon Alloy Catalysts by Carbonaceous Additives**

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**Proton exchange membrane fuel cell (PEMFC)** is a promising power source for domestic supply and automobile.

## Advantages

- High energy density.
- Compact size.
- Cleanliness.
- Low operation temperature.

## Disadvantages

- Slow cathode reaction
- $$\text{O}_2 + 4\text{H}^+ + 4\text{e}^- \rightarrow 2\text{H}_2\text{O}$$

So far, platinum has been used, but it is really expensive.

**Non-platinum catalysts are needed.**

Many types of non-Pt catalysts have been explored.  
For example...

- **Metal chalcogenide catalysts**

\*G. A. Tritsarlis, et al., *Electrochim. Acta* 56 (2011) 9783-9788.

- **Metal carbide-based catalysts**

\*A. R. Ko, et al., *Appl. Catal. Gen.* 477 (2014) 102-108.

- **N<sub>4</sub> metal complex catalysts**

\*F. Charreteur et al., *Energy Environ. Sci.* 4 (2011) 114-130.

- **Carbon alloy (CA) catalysts**

- **Nanoshell-containing carbon**

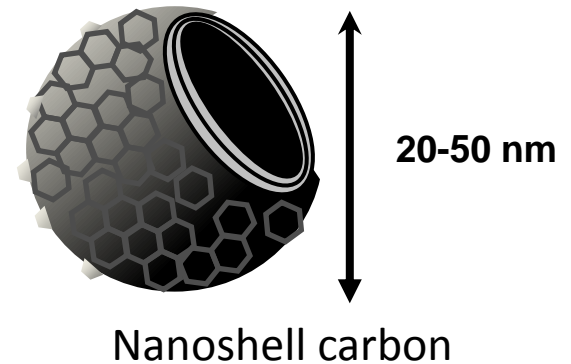
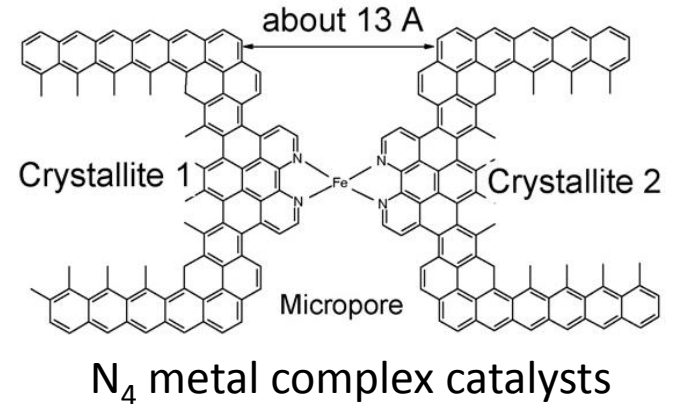
\*J. Ozaki et al., *Electrochim. Acta.* 55 (2010) 1864-1871.

- **N-doped carbon**

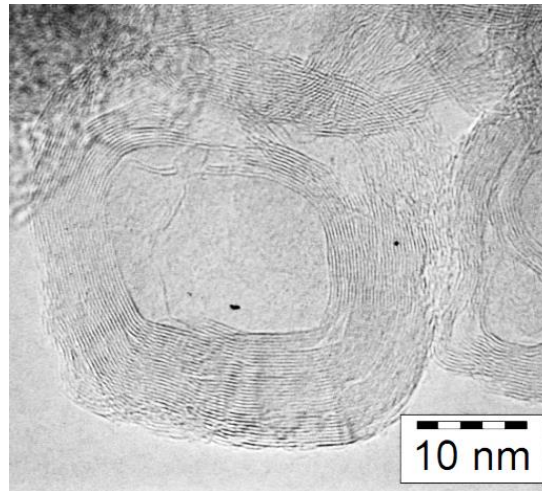
\*P. H. Matter et al., *J. Catal.* 239 (2006) 83-96.

- **BN-doped carbon**

- \*J. Ozaki et al., *Carbon* 45 (2007) 1847-1853.



# What is nanoshell-containing carbons (NSCCs)?



- **Shell-like structure (Nanoshell : NS).**
- **Formed by catalytic carbonization.**
- **Oxygen reduction reaction (ORR) activity.**

# Carbonaceous additives used

This additive was selected due to the following features of the materials.

## **Graphene oxide (GO)**

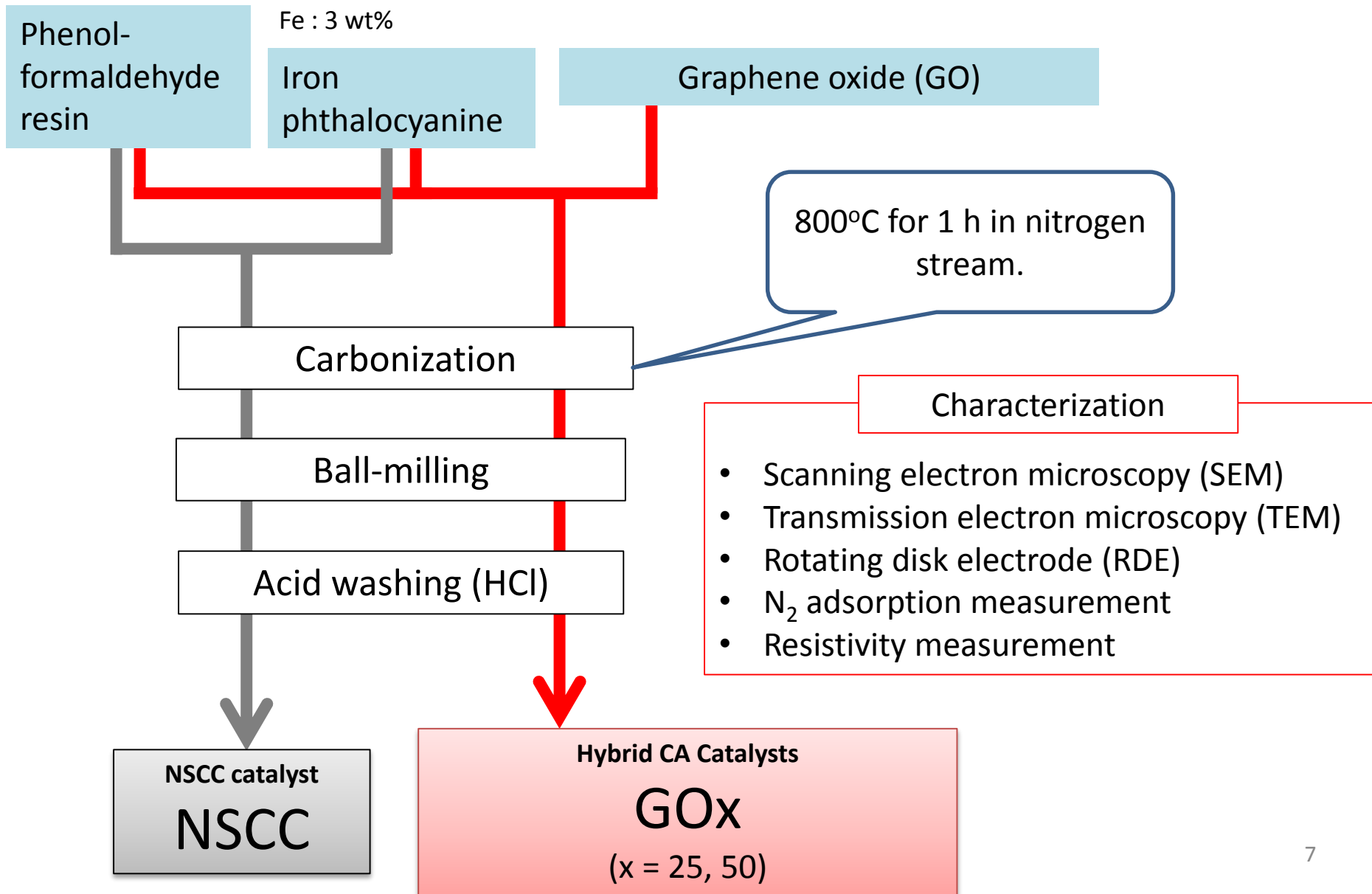
- High hydrophilic
- High conductivity  
(After pyrolysis)

# Objectives

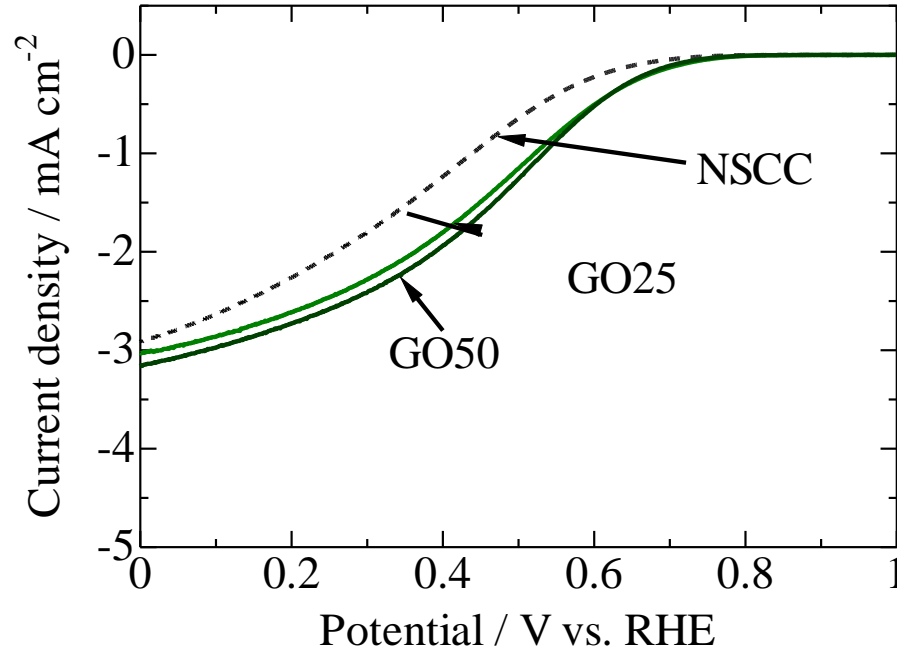
To make clear the effects of the carbon addition to a NSCC precursor on the following points :

- ( 1 ) The ORR activities of the CA catalysts.
- ( 2 ) The major factor of the ORR activity.

# Preparation process of Hybrid CA Catalysts



# ORR activity



**GO-series showed higher ORR activities than NSCC.**

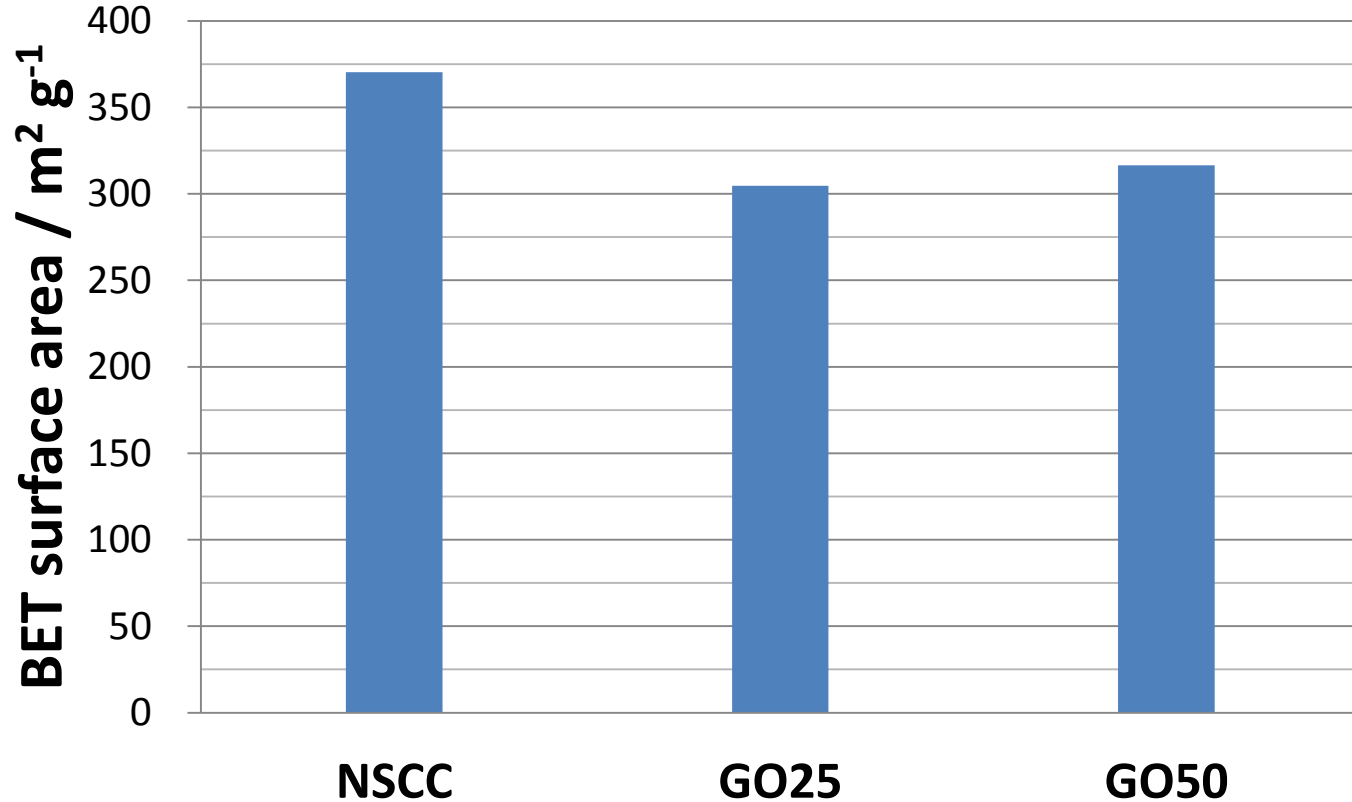


# **Possible factors to control the ORR activity**

**(1) Surface area**

**(3) Manner of contact**

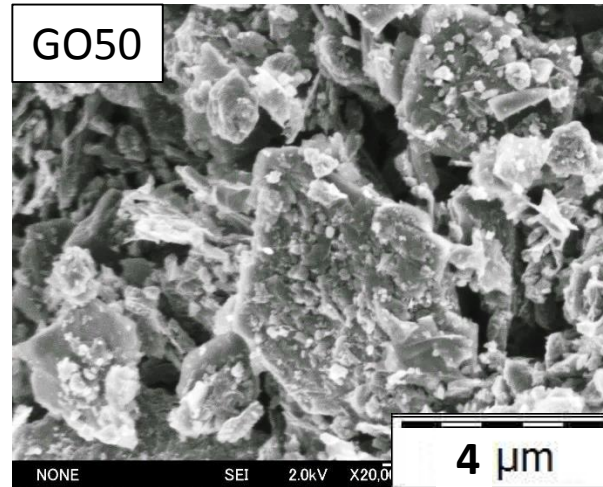
# Surface area



The BET specific surface areas was not changed by the amounts of the additives.

**BET surface areas can be a the major factor of ORR activities**

# Manner of contacts



GO-series : Flake structure derived from GO. Thick coating.

The ORR activities depended on the manner of contacts.

# Conclusion

**Investigate the influence of addition of Carbonaceous additive on the ORR activity and the structure of carbon catalysts.**

- 1. Influence of carbon addition on the ORR activity.**
  - GO enhanced ORR activities of CA catalysts.**
- 2. Influence of carbon addition on the structure.**
  - The manner of contact surface areas depended on type of additives.**